TERNARY SURFACE COMPLEX: COADSORPTION OF Cu(II), Zn(II), Cd(II) AND NITRILOTRIS(METHYLENE PHOSPHONIC) ACID ONTO BOEHMITE

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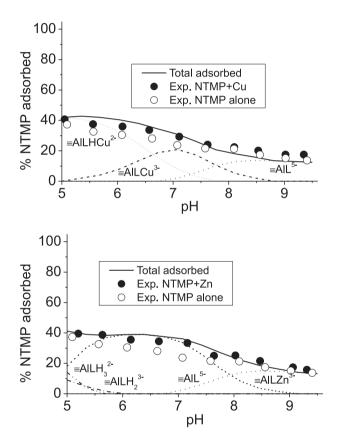


Figura 1S. Percentage NTMP 5 x 10^4 M adsorbed vs. pH in the presence of Cu(II) and Zn(II) 5 x 10^4 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO₃

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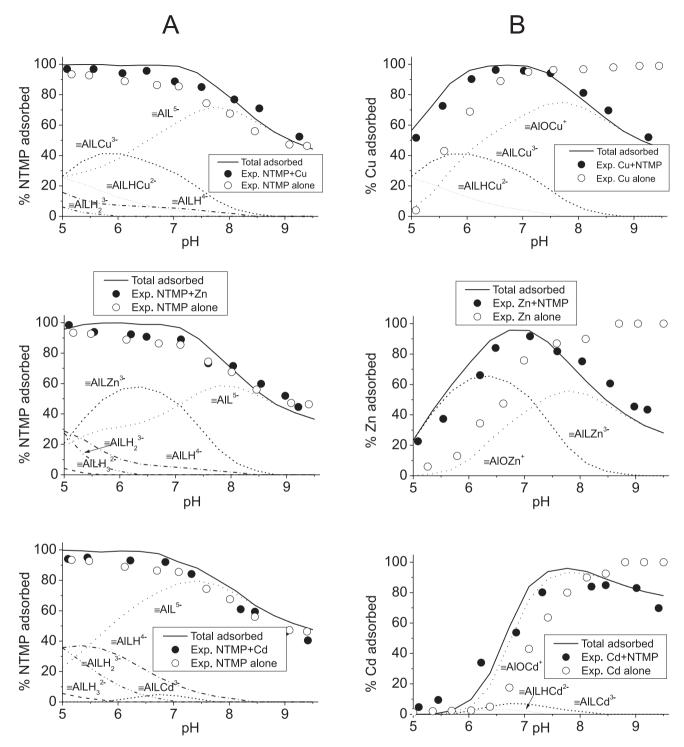


Figura 2S. A. Percentage NTMP 1 x 10^4 M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 1 x 10^4 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO₃. B. Percentage Cu(II), Zn(II) and Cd(II) 1 x 10^4 M adsorbed vs. pH in the presence of NTMP 1 x 10^4 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO₃.

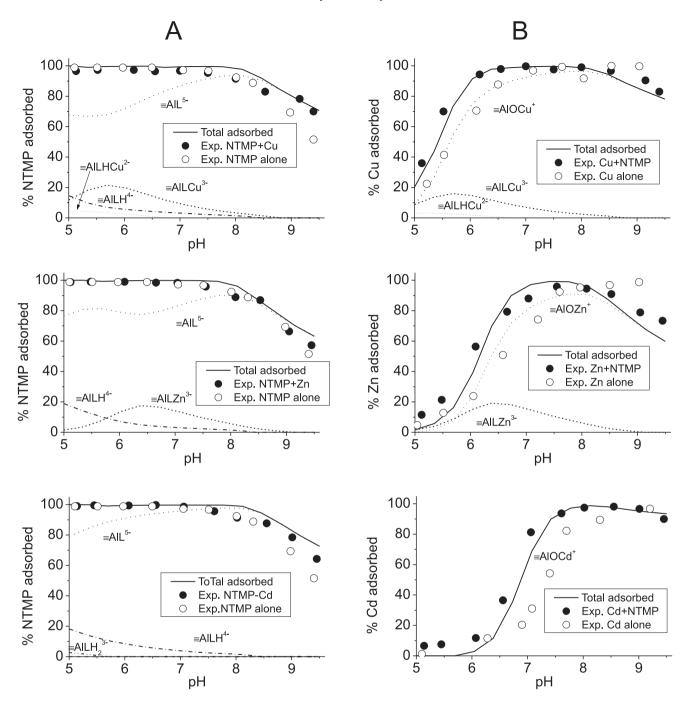


Figura 3S. A. Percentage NTMP 5 x 10^5 M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 5 x 10^5 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO₃. B. Percentage Cu(II), Zn(II) and Cd(II) 5 x 10^5 M adsorbed vs. pH in the presence of NTMP 5 x 10^5 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO₃

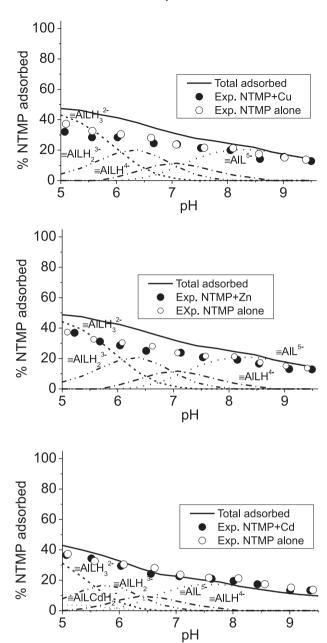


Figura 4S. Percentage NTMP 5 x 10^4 M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 1 x 10^4 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO,

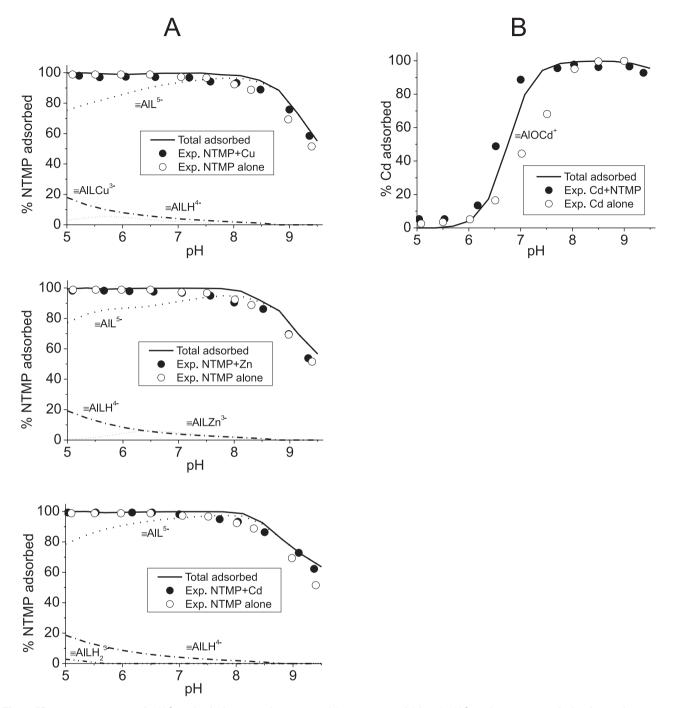


Figura 5S. A. Percentage NTMP 5 x 10-5 M adsorbed vs. pH in the presence of Cu(II), Zn(II) and Cd(II) 1 x 10-5 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO $_3$. B. Percentage Cd(II) 1 x 10-5 M adsorbed vs. pH in the presence of NTMP 5 x 10-5 M. The curves are calculated using the constants from Table 1. Conditions: 1 g/L boehmite, 0.1 M NaNO $_3$